

Supporting Information

For

Fluorescence Sensing of Ammonium and Organoammonium ions Using Tripodal Oxazoline Receptors

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Figure S1. UV absorption spectrum of oxazoline **1a**, 1.0 mM in acetonitrile.

Figure S2. UV absorption spectrum of oxazoline **1c**, 0.2 mM in acetonitrile.

Figure S3. Fluorescence emission changes of tripodal oxazoline **1c** (0.05 mM) upon addition of NH_4^+ (as ClO_4^- salt; from the top: 0.25, 0.50, 0.75, 1.00, 1.25 and 1.50 equiv. with respect to **1b**) in the presence of an excess amount (10 molar equiv.) of K^+ClO_4^- in acetonitrile following at 282-nm excitation.

Figure S4. Changes of UV absorption spectra of **1a** upon addition of NH_4^+ (as ClO_4^- salt) in acetonitrile at 25 °C.

Figure S5. Fluorescence emission changes of tripodal oxazoline **1a** (1.0 mM) upon addition of NH_4^+ (as ClO_4^- salt) in the presence of an excess amount (10 molar equiv) of K^+ClO_4^- in acetonitrile following at 272-nm excitation.

Figure S6. Fluorescence emission changes of tripodal oxazoline **1b** (1.0 mM) upon addition of NH_4^+ (as ClO_4^- salt)

Figure S7. Fluorescence emission changes of tripodal oxazoline **1b** (1.0 mM) upon addition of $\text{PhCH}_2\text{CH}_2\text{NH}_3^+$ (as ClO_4^- salt)

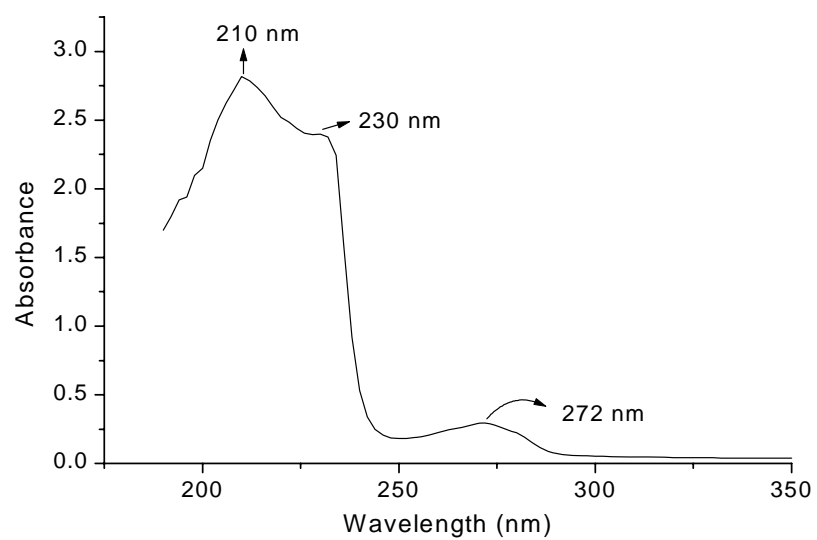


Figure S1. UV absorption spectrum of oxazoline **1a**, 1.0 mM in acetonitrile.

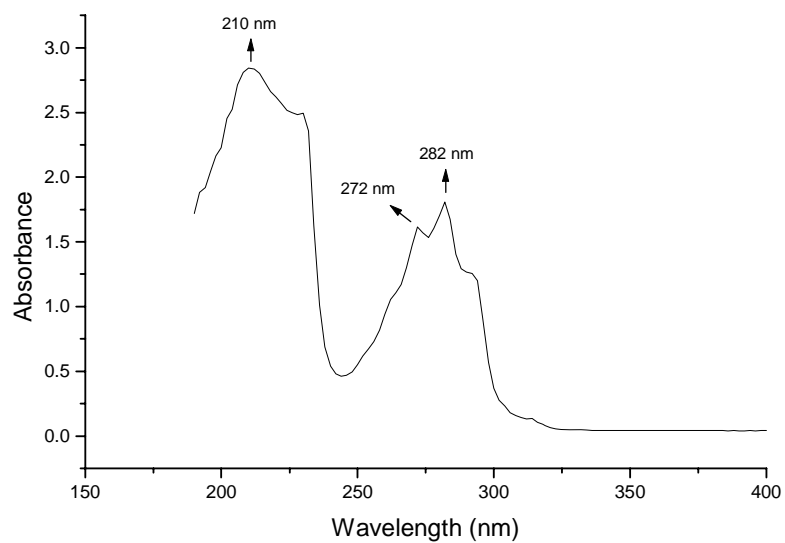


Figure S2. UV absorption spectrum of oxazoline **1c**, 0.2 mM in acetonitrile.

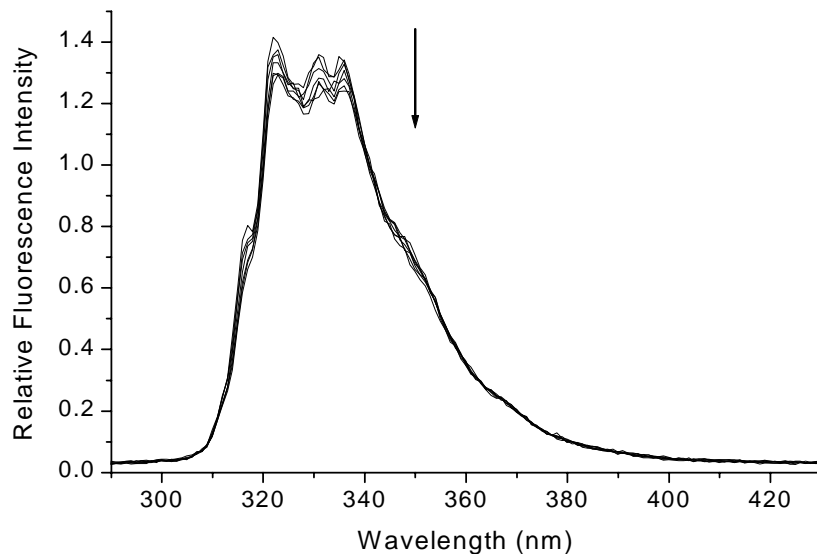


Figure S3. Fluorescence emission changes of tripodal oxazoline **1c** (0.05 mM) upon addition of NH_4^+ (as ClO_4^- salt; from the top: 0.25, 0.50, 0.75, 1.00, 1.25 and 1.50 equiv. with respect to **1c**) in the presence of an excess amount (10 molar equiv.) of K^+ClO_4^- in acetonitrile following at 282-nm excitation. Very small decrease in the intensity was observed.

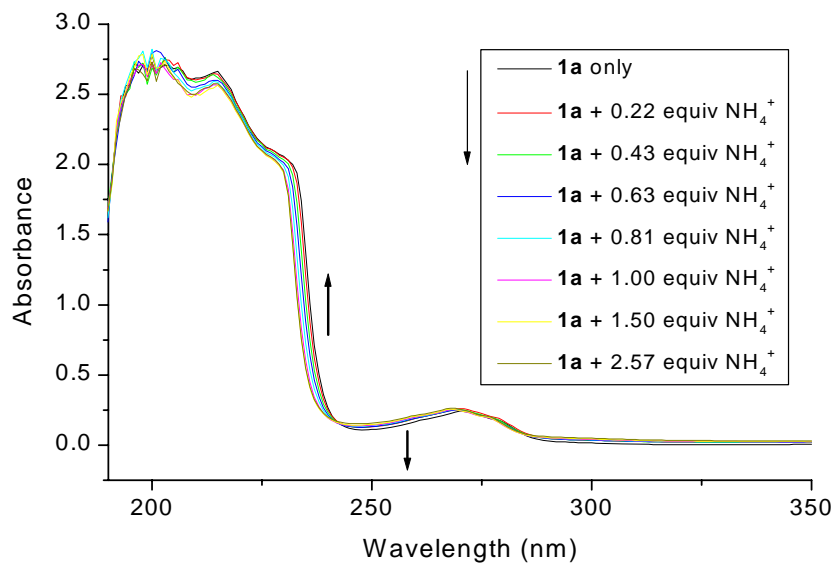


Figure S4. Changes of UV absorption spectra of **1a** upon addition of NH_4^+ (as ClO_4^- salt) in acetonitrile at 25 °C.

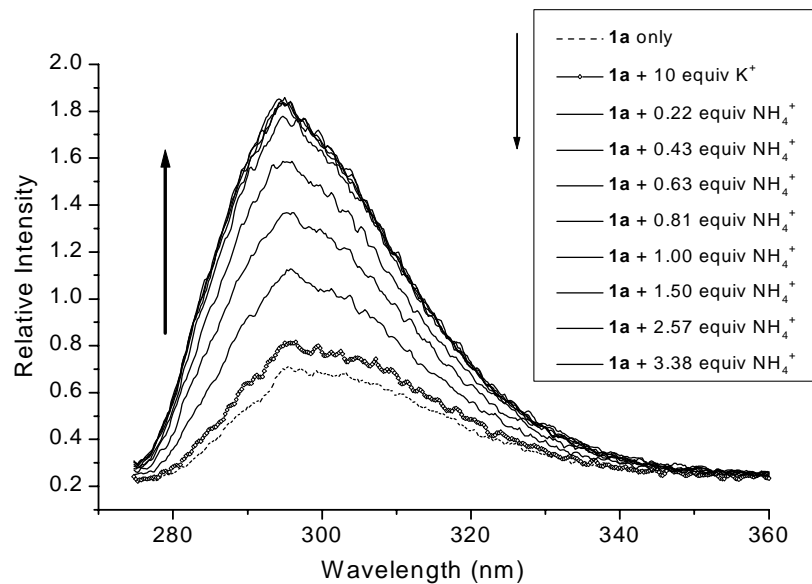


Figure S5. Fluorescence emission changes of tripodal oxazoline **1a** (1.0 mM) upon addition of NH_4^+ (as ClO_4^- salt) in the presence of an excess amount (10 molar equiv) of K^+ClO_4^- in acetonitrile following at 272-nm excitation.

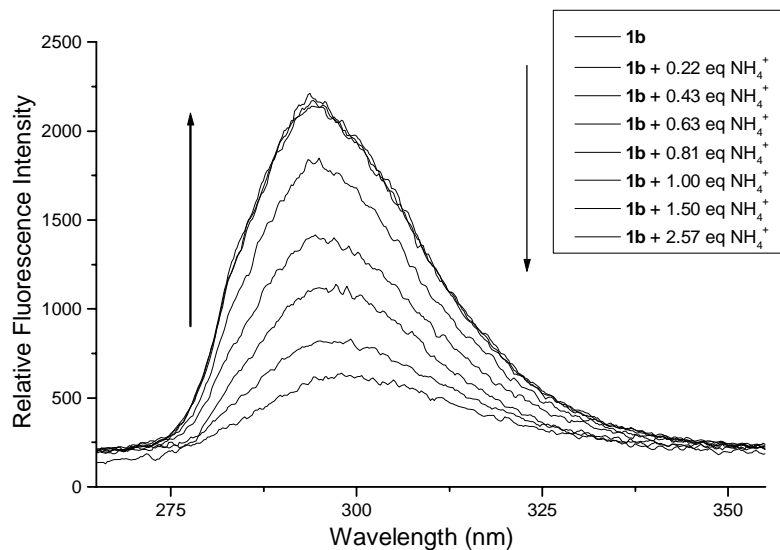


Figure S6. Fluorescence emission changes of tripodal oxazoline **1b** (1.0 mM) upon addition of NH_4^+ (as ClO_4^- salt) in acetonitrile following at 272-nm excitation.

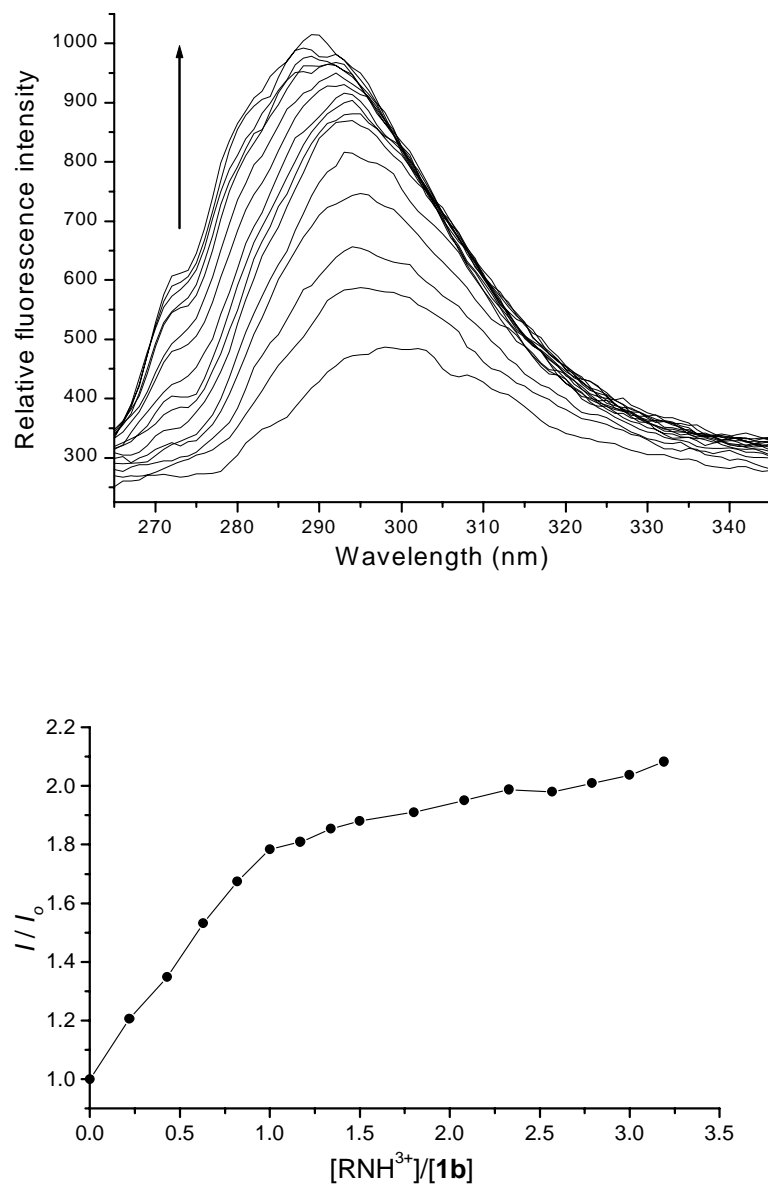


Figure S7. Fluorescence emission changes of tripodal oxazoline **1b** (1.0 mM) upon addition of PhCH₂CH₂NH₃⁺ (as ClO₄⁻ salt) in acetonitrile following at 272-nm excitation. The lower plot shows the relative fluorescence intensity depending on the molar ratio, [PhCH₂CH₂NH₃⁺]/[**1b**].